

**SULIT**



**BAHAGIAN PEPERIKSAAN DAN PENILAIAN  
JABATAN PENDIDIKAN POLITEKNIK DAN KOLEJ KOMUNITI  
KEMENTERIAN PENGAJIAN TINGGI**

**JABATAN KEJURUTERAAN PETROKIMIA**

**PEPERIKSAAN AKHIR  
SESI I : 2022 / 2023**

**DGP30122 : HEAT TRANSFER**

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**TARIKH : 13 DISEMBER 2022  
MASA : 8.30 PAGI – 10.30 PAGI (2 JAM)**

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Kertas ini mengandungi **TUJUH (7)** halaman bercetak.

Struktur (4 soalan)

Dokumen sokongan yang disertakan : Jadual, Formula

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**JANGAN BUKA KERTAS SOALANINI SEHINGGA DIARAHKAN**  
(CLO yang tertera hanya sebagai rujukan)

**SULIT**



**INSTRUCTION:**

This paper consists of **FOUR (4)** structured questions. Answer all questions.

**ARAHAN:**

*Kertas ini mengandungi **EMPAT (4)** soalan berstruktur. Jawab semua soalan.*

**QUESTION 1****SOALAN 1**

- (a) Heat spontaneously flows from a hotter area to a colder area. For example, heat is conducted from the hotplate of an electric stove to the bottom of a saucepan in contact with it.

*Haba mengalir secara spontan dari kawasan yang lebih panas ke kawasan yang lebih sejuk. Sebagai contoh, haba dijalankan dari plat panas dapur elektrik ke bahagian bawah periuk yang bersentuhan dengannya.*

- i. Give definition of convection.

*Beri definisi perolakan.*

[2 marks]

[2 markah]

CLO 1

C1

CLO 1

C1

- ii. List **TWO (2)** heat transfer mechanism other than conduction.

*Senaraikan **DUA (2)** mekanisme pemindahan haba selain konduksi.*

[4 marks]

[4 markah]

CLO 1

C2

- (b) Explain the ability of heat transfer for heat conductor and heat insulator related to thermal conductivity,  $k$ .

*Terangkan keupayaan pemindahan haba bagi konduktor haba dan penebat haba berkaitan dengan kekonduksian terma,  $k$ .*

[4 marks]

[4 markah]

- (c) A stainless-steel pot of which the thermal conductivity is  $240 \text{ W/m} \cdot ^\circ\text{C}$  has a flat bottom with a diameter of 25 cm and thickness of 0.2 cm. Heat is transferred steadily to boiling water in the pot through its bottom at a rate of 8000 W.

*Satu keluli tahan karat mempunyai kekonduksian terma  $240 \text{ W/m} \cdot ^\circ\text{C}$  berdiameter 25cm dan berketinggi 0.2 cm. Haba dipindahkan secara berterusan ke air mendidih di dalam periuk melalui permukaan bawah dengan kadar 8000W.*

CLO 2  
C3

- i. Calculate temperature of the bottom outer surface of the pot if the inner bottom surface of the pot is at  $100^{\circ}\text{C}$ .

*Kira suhu permukaan luar bawah periuk tersebut sekiranya suhu permukaan dalam periuk tersebut  $100^{\circ}\text{C}$ .*

[7 marks]  
[7 markah]

CLO 2  
C3

- ii. Aluminium pot has the same size as the stainless-steel pot. Calculate the heat transfer rate by conduction through its bottom if the thermal conductivity of aluminium is  $300 \text{ W/m} \cdot ^{\circ}\text{C}$ . Use the answer in (c)(i) as the outer surface temperature and the inner bottom surface of the pot is at  $100^{\circ}\text{C}$ .

*Periuk aluminium mempunyai saiz yang sama dengan periuk keluli tahan karat. Kira kadar pemindahan haba menerusi kunduksi melalui bahagian bawah sekiranya kekonduksian terma bagi aluminium ialah  $300 \text{ W/m} \cdot ^{\circ}\text{C}$ . Guna jawapan di (c)(i) sebagai suhu permukaan luar dan suhu permukaan dalam bagi bawah periuk ialah  $100^{\circ}\text{C}$ .*

[8 marks]  
[8 markah]

## QUESTION 2

### SOALAN 2

CLO 1  
C1

- (a) Define internal flow and external flow of pipe.

*Takrifkan aliran dalaman dan aliran luaran paip.*

[4 marks]  
[4 markah]

- (b) Engine oil with kinematic viscosity of  $2.046 \times 10^{-5} \text{ m}^2/\text{s}$  and  $840 \text{ kg/m}^3$  density flow over a  $2\text{m} \times 2\text{m}$  flat plate with a velocity of  $2 \text{ m/s}$ . The convective heat transfer coefficient is  $4.5 \text{ W}/^{\circ}\text{C}$ .

*Minyak enjin dengan kelikatan kinematik  $2.046 \times 10^{-5} \text{ m}^2/\text{s}$  and  $840 \text{ kg/m}^3$  tumpat mengalir atas  $2\text{m} \times 2\text{m}$  plat ratadengan had laju  $2 \text{ m/s}$ . Purata pekali pemindahan haba ialah  $4.5 \text{ W}/^{\circ}\text{C}$ .*

CLO 1

C2

- i. Discuss the flow condition of engine oil (laminar or turbulent) based on the calculated Reynold Number.

*Bincangkan kondisi pergerakan minyak enjin (laminar atau turbulen) berdasarkan Nombor Reynold yang dikira.*

[4 marks]

[4 markah]

CLO 1

C2

- ii. This process is required to remove 150W of heat from the surface. The surface temperature of the flat plate,  $T_s$  is 35°C and the engine oil flows at temperature 25°C. Report whether this process is achievable or not.

*Proses ini perlu menyingkirkan 150W haba daripada permukaan. Suhu permukaan bagi plat rata,  $T_s$  ialah 35°C dan minyak enjin mengalir pada suhu 25°C. Laporkan sama ada proses ini boleh dicapai atau tidak.*

[3 marks]

[3 markah]

- (c) In a winter season, a 0.2m diameter of a process line pipe with external surface temperature 150°C passes through one open area that is not protected against the surrounding cold temperature. The wind is blowing across the cylindrical pipe at a velocity of 10m/s and the temperature is -10°C (1 atm pressure).

*Pada musim sejuk, satu paip laluan proses berdiameter 0.2m dengan suhu permukaan luar 150°C melalui satu kawasan terbuka yang tidak dilindungi daripada suhu luar yang sejuk. Angin tersebut bertemu menyeberangi paip silinder itu pada had laju 10 m/s dan suhu -10°C (1atm tekanan).*

CLO 2

C3

- i. Calculate the Nusselt Number,  $N_{Nu}$  for this case.

*Kira Nombor Nusselt,  $N_{Nu}$  untuk kes ini.*

[7 marks]

[7 markah]

CLO 2

C3

- ii. Calculate convective heat transfer rate across that cylinder pipe if the length of pipe is 10m.

*Kira kadar pemindahan haba secara perolakan menyeberangi paip silinder tersebut sekiranya panjang paip ialah 10m.*

[7 marks]

[7 markah]

**QUESTION 3**  
**SOALAN 3**

- (a) Radiation is the emission or transmission of energy in the form of waves or particles through space or through a material medium.

*Radiasi adalah pelepasan atau penghantaran tenaga dalam bentuk gelombang atau zarah melalui ruang atau melalui medium bahan.*

CLO 1  
C2

- i. Explain total blackbody emissive power.

*Jelaskan kuasa pancaran penuh jasad hitam.*

[3 marks]

[3 markah]

CLO 1  
C2

- ii. Explain spectral blackbody emissive power.

*Jelaskan kuasa pancaran spectrum jasad hitam.*

[3 marks]

[3 markah]

- (b) Imagine a spherical ball with a diameter of 0.5m at the temperature of 1000 K is suspended in air. Assume the ball is a blackbody.

*Bayangkan satu bola sfera dengan diameter 0.5m pada suhu 1000K terapung di udara. Anggap bola tersebut sebuah jasad hitam.*

CLO 2  
C3

- i. Calculate the blackbody maximum rate of radiation emitted by the ball.

*Kira kadar maksimum radiasi jasad hitam yang dipancarkan oleh bola tersebut.*

[4 marks]

[4 markah]

CLO 2  
C3

- ii. Calculate spectral blackbody emissive power at a wavelength of  $5\mu\text{m}$  and  $7\mu\text{m}$  at a temperature of 1000K.

*Kira kuasa pancaran spectrum jasad hitam pada panjang gelombang  $5\mu\text{m}$  dan  $7\mu\text{m}$  pada suhu 1000K.*

[7 marks]

[7 markah]

CLO 2

C3

- (c) The temperature of filament of an incandescent lightbulb is 3500 K. Assume the filament to be a blackbody, calculate the fraction of the radiant energy emitted by the filament that falls in the visible range. The visible range of the electromagnetic spectrum extends from  $\lambda_1 = 0.4 \mu\text{m}$  to  $\lambda_2 = 0.8 \mu\text{m}$ .

*Suhu filamen mentol pijar ialah 3500 K. Andaikan filamen itu ialah jasad hitam, kirakan pecahan tenaga sinaran yang dipancarkan oleh filamen yang jatuh dalam julat yang boleh dilihat. Julat spektrum elektromagnet yang boleh dilihat menjangkau dari  $\lambda_1 = 0.4 \mu\text{m}$  hingga  $\lambda_2 = 0.8 \mu\text{m}$ .*

[8 marks]

[8 markah]

**QUESTION 4****SOALAN 4**

CLO 1

C1

- (a) Heat Exchanger is one of the critical equipment in the industry especially when involving chemicals, liquid and others. List **FOUR (4)** types of Heat Exchanger.

*Penukar haba adalah salah satu peralatan kritikal dalam industri terutamanya apabila melibatkan bahan kimia, cecair dan lain-lain. Senaraikan **EMPAT (4)** jenis penukar haba.*

[4 marks]

[4 markah]

- (b) Shell and tubes heat exchanger is the common types of heat exchanger found in petrochemical industry.

*Penukar haba cengkerang dan tiub adalah penukar haba yang biasa ditemui di industry petrokimia.*

CLO 1

C2

- i. Explain **TWO (2)** properties of shell and tubes heat exchanger.

*Jelaskan **DUA (2)** ciri-ciri penukar haba cengkerang dan tiub.*

[5 marks]

[5 markah]



CLO 1

C2

- ii. Explain the fouling phenomenon in heat exchanger.

*Terangkan fenomena kekotoran dalam penukar haba.*

[5 marks]

[5 markah]

- (c) Hot oil is to be cooled in a double pipe counter-flow heat exchanger. Cold water flows through the tube while the oil flows through the shell. The tube diameter is 2cm with negligible thickness while the shell diameter is 4cm.

*Minyak panas hendaklah disejukkan dalam penukar haba paip berganda yang beraliran balas. Air sejuk mengalir melalui tiub manakala minyak mengalir melalui cangkerang. Diameter tiub ialah 2cm dengan mengabaikan ketebalan manakala diameter cangkerang ialah 4cm.*

CLO 2

C3

- i. Draw the double pipe counter-flow heat exchanger with a complete label of hot oil inlet, hot oil outlet, cold water inlet, cold water outlet, shell and tube.

*Lukis penukar haba aliran balas paip berganda dengan label lengkap salur masuk minyak panas, salur keluar minyak panas, salur masuk air sejuk, salur keluar air sejuk, cengkerang dan tiub.*

[5 marks]

[5 markah]

CLO 2

C3

- ii. The Nusselt Number and thermal conductivity for the flowing cold water is 240 and 0.5 W/m.K. The convective heat transfer coefficient of oil is 75.5W/m<sup>2</sup>.K. Calculate the overall convective heat transfer coefficient, U for the double pipe counter-flow heat exchanger.

*Nombor Nusselt dan kekonduksian terma bagi air sejuk yang mengalir ialah 240 dan 0.5 W/m.K. Pekali pemindahan haba perolakan minyak ialah 75.5W/m<sup>2</sup>.K. Kirakan pekali pemindahan haba perolakan keseluruhan, U untuk penukar haba aliran balas paip berganda.*

[6 marks]

[6 markah]

**SOALAN TAMAT**



## HEAT TRANSFER FORMULA

### Conduction

$$Q_{\text{conduction}} = \frac{T_1 - T_2}{R}$$

$$Q_{\text{conduction,plate}} = kA \frac{T_1 - T_2}{X}$$

$$\text{Conduction Resistance (Plane)}, R_{\text{wall}} = \frac{X}{kA}$$

$$\text{Conduction Resistance (Cylinder)}, R_{\text{cyl}} = \frac{\ln(r_2/r_1)}{2\pi Lk}$$

$$\text{Conduction Resistance (Sphere)}, R_{\text{sphere}} = \frac{r_2 - r_1}{4\pi r_1 r_2 k}$$

$$\text{Thermal Resistance Concept}, Q = \frac{\Delta T}{R_{\text{total}}}$$

Series Thermal Resistance,  $R_{\text{total}} = R_1 + R_2 + R_3 + \dots$

$$\text{Parallel Thermal Resistance}, \frac{1}{R_{\text{total}}} = \frac{1}{R_1} + \frac{1}{R_2} + \dots$$

### Convection

$$\dot{Q}_{\text{conv}} = hA_s(T_s - T_\infty)$$

$$\text{Convection Resistance, } R_{\text{conv}} = \frac{1}{Ah}$$

$$\text{Film, temperature, } T_f = \frac{T_s + T_\infty}{2}$$

$$\text{Nusselt number, } N_{\text{Nu}} = \frac{hL}{k} = \frac{hD}{k}$$

$$\text{Reynold Number for plate, } N_{\text{Re}} = \frac{\rho v L}{\mu}$$

$$\text{Reynold Number for plate, } N_{\text{Re}} = \frac{vL}{v}$$

$$\text{Reynold Number for cylinder/sphere, } N_{\text{Re}} = \frac{\rho v D}{\mu}$$

$$\text{Reynold Number for cylinder/sphere, } N_{\text{Re}} = \frac{vD}{v}$$

### **Parallel flow over a flat plate**

**Laminar flow** (For Reynold Number,  $N_{\text{Re}} < 5 \times 10^5$ ):

$$N_{\text{Nu}} = 0.664 N_{\text{Re}}^{0.5} Pr^{1/3}$$

**Turbulence flow** (For Reynold Number,  $5 \times 10^5 \leq N_{\text{Re}} \leq 10^7$  and  $0.6 \leq Pr \leq 60$ ):

$$N_{\text{Nu}} = 0.037 N_{\text{Re}}^{0.8} Pr^{1/3}$$

### **Flow across cylinder**

$$N_{\text{Nu}} = 0.3 + \frac{0.62 N_{\text{Re}}^{1/2} Pr^{1/3}}{[1 + (\frac{N_{\text{Re}}}{282000})^{5/8}]^{4/5}} [1 + \left(\frac{0.4}{Pr}\right)^3]^{1/4}$$

**Note:** For  $(N_{\text{Re}} \times Pr > 0.2)$ , Fluid properties at  $T_f$ .

### Flow across sphere

$$N_{Nu} = 2 + \left[ 0.4N_{Re}^{\frac{1}{2}} + 0.06N_{Re}^{\frac{2}{3}} \right] Pr^{0.4} \left( \frac{\mu_{\infty}}{\mu_s} \right)^{1/4}$$

**Note:** Fluid properties is referring to the surrounding temperature,  $T_{\infty}$  except  $\mu_s$  is referred to  $T_s$ .

### Flow across tube banks

$$A_1 = S_T L$$

$$A_T = (S_T - D)L$$

$$A_D = (S_D - D)L$$

#### In line flow direction:

$$V_{max} = \frac{S_T}{S_T - D} (V)$$

#### Staggered flow direction:

$$\text{If } S_D > \frac{(S_T + D)}{2} :$$

$$V_{max} = \frac{S_T}{S_T - D} V$$

$$\text{If } S_D < \frac{(S_T + D)}{2} :$$

$$V_{max} = \frac{S_T}{2(S_D - D)} V$$

$$N_{Re} = \frac{\rho V_{max} D}{\mu} = \frac{V_{max} D}{v}$$

$$N_{Nu,new} = F \cdot N_{Nu}$$

$$A_s = N\pi D L$$

$$N = N_L \times N_T$$

$$Q = A_s h \Delta T_{in}$$

### Nusselt Number correlation in tube banks:

Arrangement	Range of $Re_D$	Correlation
In-line	0–100	$Nu_D = 0.9 Re_D^{0.4} Pr^{0.36} (Pr/Pr_s)^{0.25}$
	100–1000	$Nu_D = 0.52 Re_D^{0.5} Pr^{0.36} (Pr/Pr_s)^{0.25}$
	$1000–2 \times 10^5$	$Nu_D = 0.27 Re_D^{0.63} Pr^{0.36} (Pr/Pr_s)^{0.25}$
	$2 \times 10^5–2 \times 10^6$	$Nu_D = 0.033 Re_D^{0.8} Pr^{0.4} (Pr/Pr_s)^{0.25}$
Staggered	0–500	$Nu_D = 1.04 Re_D^{0.4} Pr^{0.36} (Pr/Pr_s)^{0.25}$
	500–1000	$Nu_D = 0.71 Re_D^{0.5} Pr^{0.36} (Pr/Pr_s)^{0.25}$
	$1000–2 \times 10^5$	$Nu_D = 0.35(S_t/S_L)^{0.2} Re_D^{0.6} Pr^{0.36} (Pr/Pr_s)^{0.25}$
	$2 \times 10^5–2 \times 10^6$	$Nu_D = 0.031(S_t/S_L)^{0.2} Re_D^{0.8} Pr^{0.36} (Pr/Pr_s)^{0.25}$

\*All properties except  $Pr_s$  are to be evaluated at the arithmetic mean of the inlet and outlet temperatures of the fluid ( $Pr_s$  is to be evaluated at  $T_s$ ).

### Nusselt number correction factor calculation:

Correction factor  $F$  to be used in  $Nu_{D,N_L} = FNu_D$  for  $N_L < 16$  and  $Re_D > 1000$  (from Zukauskas, Ref 15, 1987).

$N_L$	1	2	3	4	5	7	10	13
In-line	0.70	0.80	0.86	0.90	0.93	0.96	0.98	0.99
Staggered	0.64	0.76	0.84	0.89	0.93	0.96	0.98	0.99

### Internal Flow Convection (Cylinder)

#### For laminar flow:

$$T_{ave\ bulk} = \frac{T_{bi} + T_{bo}}{2}$$

$$Nu_{Nu} = 1.86(N_{Re}N_{Pr}\frac{D}{L})^{\frac{1}{3}}(\frac{\mu_b}{\mu_w})^{0.14}$$

$$\dot{Q}_{conv} = h_a A_s \Delta T_a$$

$$\Delta T_a = \frac{(T_w - T_{bi}) + (T_w - T_{bo})}{2}$$

**For turbulent flow:**

$$N_{Nu} = 0.027 N_{Re}^{0.8} Pr^{\frac{1}{3}} \left( \frac{\mu_b}{\mu_w} \right)^{0.14}$$

**Note:** For  $N_{Re} > 6000$ ,  $0.7 < Pr < 16000$  and  $L/D > 60$

$$\dot{Q}_{conv} = h_L A_s \Delta T_{lm}$$

$$\Delta T_{lm} = \frac{(T_w - T_{bi}) - (T_w - T_{bo})}{\ln(\frac{T_w - T_{bi}}{T_w - T_{bo}})}$$

## Radiation

$$Q_{\text{rad}} = \epsilon \sigma A_s (T_s^4 - T_\infty^4), \sigma = 5.670 \times 10^{-8} \text{ W/m}^2 \cdot \text{K}^4$$

$$E_{b\lambda}(\lambda, T) = \frac{C_1}{\lambda^5 [\exp(C_2/\lambda T) - 1]} \text{ W/m}^2 \cdot \mu\text{m}$$

$$C_1 = 3.742 \times 10^8 \text{ W} \cdot \mu\text{m}^4 / \text{m}^2$$

$$C_2 = 1.439 \times 10^4 \mu\text{m} \cdot \text{K}$$

**Fraction of radiation:**

$$f_{\lambda_1-\lambda_2}(T) = f_{\lambda_2}(T) - f_{\lambda_1}(T)$$

**Wien's displacement law:**

$$\lambda T = 2897.8 \mu\text{m} \cdot \text{K}$$

### Heat Exchanger

$$R_{\text{total}} = R_i + R_{\text{wall}} + R_o = \frac{1}{h_i A_i} + \frac{(D_o/D_i)}{2\pi k L} + \frac{1}{h_o A_o}$$

$$R_{\text{wall}} = \frac{(D_o/D_i)}{2\pi k L}$$

$$Q = \frac{\Delta T}{R} = UA\Delta T_{LM} = U_i A_i \Delta T_{LM} = U_o A_o \Delta T_{LM}$$

$$\frac{1}{U_o A_s} = \frac{1}{U_i A_i} = \frac{1}{U_o A_o} = R_{\text{total}}$$

$$T_{\text{ave}} = \frac{T_{bi} + T_{bo}}{2}$$

$$\frac{1}{U} \approx \frac{1}{h_i} + \frac{1}{h_o}$$

**Fouling factor:**

$$R_{\text{fouling}} = \frac{R_{\text{factor}}}{A}$$

**Shell flow (Laminar):**

<b>D<sub>i</sub> / D<sub>o</sub></b>	<b>N<sub>Nu</sub></b>
0	0
0.05	17.46
0.1	11.56
0.25	7.37
0.5	5.74
1.0	4.86

**Shell flow (Turbulent):**

$$N_{Nu} = 0.3 + \frac{0.62 N_{Re}^{1/2} Pr^{1/3}}{\left[1 + \left(\frac{0.4}{Pr}\right)^{\frac{2}{3}}\right]^{1/4}} \left[1 + \left(\frac{N_{Re}}{282000}\right)^{\frac{5}{8}}\right]^{4/5}$$

**Tube flow (Laminar):**

$$N_{Nu} = 1.86 \left( N_{Re} N_{Pr} \frac{D}{L} \right)^{\frac{1}{3}} \left( \frac{\mu_b}{\mu_w} \right)^{0.14}$$

**Tube flow (Turbulent):**

$$Nu = 0.023 Re^{0.8} Pr^{0.4}$$

**Table of Unit Conversion**

<b>Quantity</b>	<b>Equivalent Values</b>
<b>Mass</b>	1 kg = 1000 g = 0.001 metric ton = 2.20462 lb <sub>m</sub> = 35.27392 oz 1 lb <sub>m</sub> = 16 oz = 5 X 10 <sup>-4</sup> ton = 453.593 g = 0.453593 kg
<b>Length</b>	1 m = 100 cm = 1000 mm = 10 <sup>6</sup> microns ( μm ) = 10 <sup>10</sup> angstroms ( Å ) = 39.37 in. = 3.2808 ft = 1.0936 yd = 0.0006214 mile
<b>Volume</b>	1 m <sup>3</sup> = 1000 liters = 10 <sup>6</sup> cm <sup>3</sup> = 10 <sup>6</sup> ml = 35.3145 ft <sup>3</sup> = 220.83 imperial gallons = 264.17 gal = 1056.68 qt 1 ft <sup>3</sup> = 1728 in <sup>3</sup> = 7.4805 gal = 0.028317 m <sup>3</sup> = 28.317 liters = 28 317 cm <sup>3</sup>
<b>Force</b>	1 N = 1 kg.m/s <sup>2</sup> = 10 <sup>5</sup> dynes = 10 <sup>5</sup> g.cm/s <sup>2</sup> = 0.22481 lb <sub>f</sub> 1 lb <sub>f</sub> = 32.174 lb <sub>m</sub> .ft/s <sup>2</sup> = 4.4482 N = 4.4482 X 10 <sup>4</sup> dynes
<b>Pressure</b>	1 atm = 1.01325 x 10 <sup>5</sup> N/m <sup>2</sup> (Pa) = 101.325 kPa = 1.01325 bars = 1.01325 x 10 <sup>6</sup> dynes/cm <sup>2</sup> = 760 mm Hg at 0 °C (torr) = 10.333 m H <sub>2</sub> O at 4 °C = 14.696lb <sub>f</sub> /in <sup>2</sup> (psi) = 33.9 ft H <sub>2</sub> O at 4 °C = 29.921 in Hg at 0 °C
<b>Energy</b>	1 J = 1 N.m = 10 <sup>7</sup> ergs = 10 <sup>7</sup> dyne.cm = 2.778 x 10 <sup>-7</sup> kW.h = 0.23901 cal = 0.7376 ft-lb <sub>f</sub> = 9.486 x 10 <sup>-4</sup> Btu
<b>Power</b>	1 W = 1J/s = 0.23901 cal/s = 0.7376 ft.lb <sub>f</sub> /s = 9.468 x 10 <sup>-4</sup> Btu/s = 1.341 x 10 <sup>-3</sup> hp

Blackbody radiation functions  $f_\lambda$

$\lambda T$ , $\mu\text{m} \cdot \text{K}$	$f_\lambda$	$\lambda T$ , $\mu\text{m} \cdot \text{K}$	$f_\lambda$
200	0.000000	6200	0.754140
400	0.000000	6400	0.769234
600	0.000000	6600	0.783199
800	0.000016	6800	0.796129
1000	0.000321	7000	0.808109
1200	0.002134	7200	0.819217
1400	0.007790	7400	0.829527
1600	0.019718	7600	0.839102
1800	0.039341	7800	0.848005
2000	0.066728	8000	0.856288
2200	0.100888	8500	0.874608
2400	0.140256	9000	0.890029
2600	0.183120	9500	0.903085
2800	0.227897	10,000	0.914199
3000	0.273232	10,500	0.923710
3200	0.318102	11,000	0.931890
3400	0.361735	11,500	0.939959
3600	0.403607	12,000	0.945098
3800	0.443382	13,000	0.955139
4000	0.480877	14,000	0.962898
4200	0.516014	15,000	0.969981
4400	0.548796	16,000	0.973814
4600	0.579280	18,000	0.980860
4800	0.607559	20,000	0.985602
5000	0.633747	25,000	0.992215
5200	0.658970	30,000	0.995340
5400	0.680360	40,000	0.997967
5600	0.701046	50,000	0.998953
5800	0.720158	75,000	0.999713
6000	0.737818	100,000	0.999905

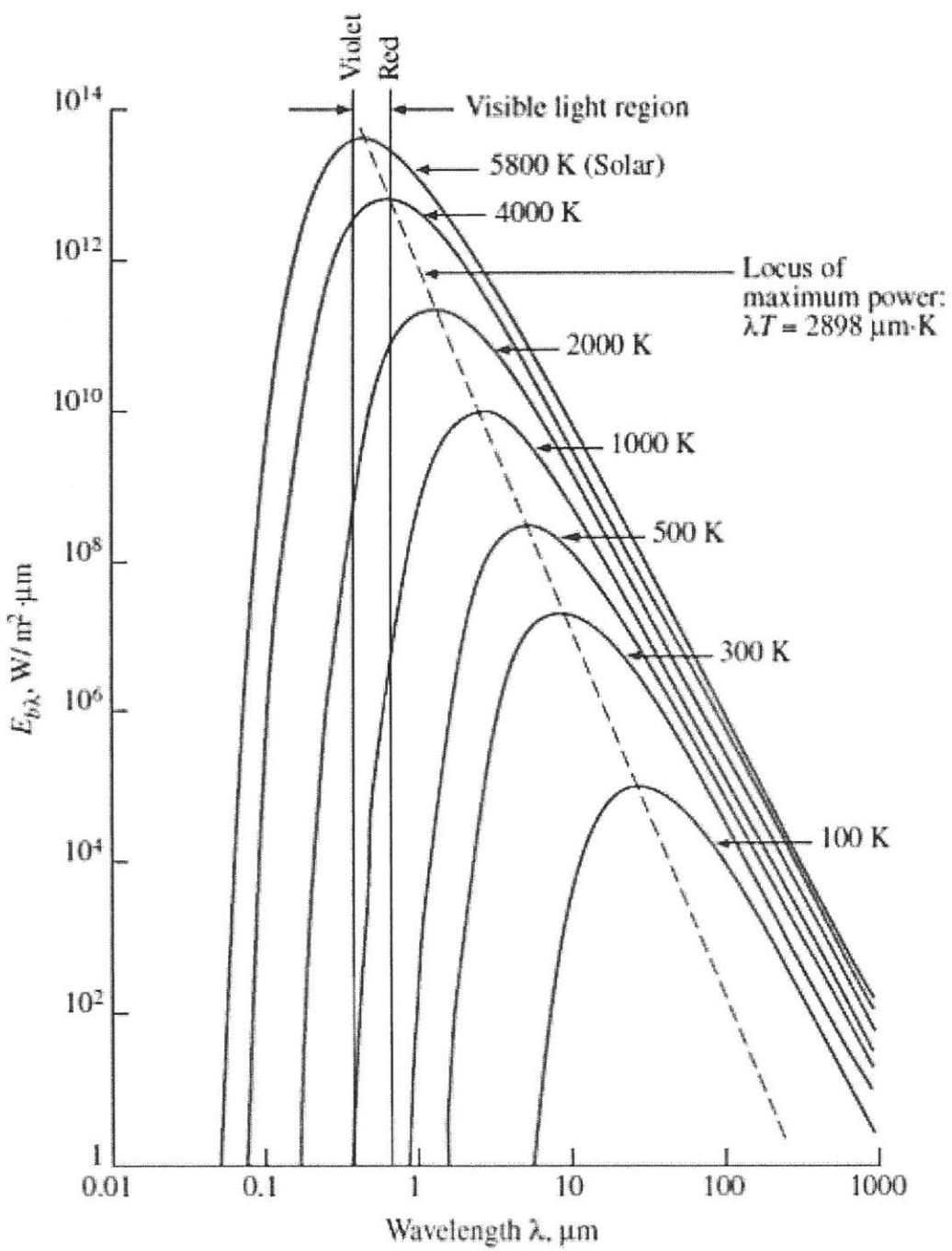




TABLE A-9

Properties of saturated water

Temp. <i>T</i> , °C	Saturation Pressure <i>P</i> <sub>sat</sub> , kPa	Density <i>ρ</i> , kg/m <sup>3</sup>		Enthalpy of Vaporization <i>h</i> <sub>fg</sub> , kJ/kg		Specific Heat <i>c</i> <sub>p</sub> , J/kg·K		Thermal Conductivity <i>k</i> , W/m·K		Dynamic Viscosity <i>μ</i> , kg/m·s		Prandtl Number Pr		Volume Expansion Coefficient <i>β</i> , 1/K	
		Liquid	Vapor	Liquid	Vapor	Liquid	Vapor	Liquid	Vapor	Liquid	Vapor	Liquid	Vapor	Liquid	Vapor
0.01	0.6113	999.8	0.0048	2501	4217	1854	0.561	0.0171	$1.792 \times 10^{-3}$	$0.922 \times 10^{-5}$	13.5	1.00	$-0.068 \times 10^{-3}$		
5	0.8721	999.9	0.0068	2490	4205	1857	0.571	0.0173	$1.519 \times 10^{-3}$	$0.934 \times 10^{-5}$	11.2	1.00	$0.015 \times 10^{-3}$		
10	1.2276	999.7	0.0094	2478	4194	1862	0.580	0.0176	$1.307 \times 10^{-3}$	$0.946 \times 10^{-5}$	9.45	1.00	$0.733 \times 10^{-3}$		
15	1.7051	999.1	0.0128	2466	4185	1863	0.589	0.0179	$1.138 \times 10^{-3}$	$0.959 \times 10^{-5}$	8.09	1.00	$0.138 \times 10^{-3}$		
20	2.339	998.0	0.0173	2454	4182	1867	0.598	0.0182	$1.002 \times 10^{-3}$	$0.973 \times 10^{-5}$	7.01	1.00	$0.195 \times 10^{-3}$		
25	3.169	997.0	0.0231	2442	4180	1870	0.607	0.0186	$0.891 \times 10^{-3}$	$0.987 \times 10^{-5}$	6.14	1.00	$0.247 \times 10^{-3}$		
30	4.246	996.0	0.0304	2431	4178	1875	0.615	0.0189	$0.798 \times 10^{-3}$	$1.001 \times 10^{-5}$	5.42	1.00	$0.294 \times 10^{-3}$		
35	5.628	994.0	0.0397	2419	4178	1880	0.623	0.0192	$0.720 \times 10^{-3}$	$1.016 \times 10^{-5}$	4.83	1.00	$0.337 \times 10^{-3}$		
40	7.384	992.1	0.0512	2407	4179	1885	0.631	0.0196	$0.653 \times 10^{-3}$	$1.031 \times 10^{-5}$	4.32	1.00	$0.377 \times 10^{-3}$		
45	9.593	990.1	0.0655	2395	4180	1892	0.637	0.0200	$0.596 \times 10^{-3}$	$1.046 \times 10^{-5}$	3.91	1.00	$0.415 \times 10^{-3}$		
50	12.35	988.1	0.0831	2383	4181	1900	0.644	0.0204	$0.547 \times 10^{-3}$	$1.062 \times 10^{-5}$	3.55	1.00	$0.451 \times 10^{-3}$		
5	15.76	985.2	0.1045	2371	4183	1908	0.649	0.0208	$0.504 \times 10^{-3}$	$1.077 \times 10^{-5}$	3.25	1.00	$0.484 \times 10^{-3}$		
60	19.94	983.3	0.1304	2359	4185	1916	0.654	0.0212	$0.467 \times 10^{-3}$	$1.093 \times 10^{-5}$	2.99	1.00	$0.517 \times 10^{-3}$		
65	25.03	980.4	0.1614	2346	4187	1926	0.659	0.0216	$0.433 \times 10^{-3}$	$1.110 \times 10^{-5}$	2.75	1.00	$0.548 \times 10^{-3}$		
70	31.19	977.5	0.1983	2334	4190	1936	0.663	0.0221	$0.404 \times 10^{-3}$	$1.126 \times 10^{-5}$	2.55	1.00	$0.578 \times 10^{-3}$		
75	38.58	974.7	0.2421	2321	4193	1948	0.667	0.0225	$0.378 \times 10^{-3}$	$1.142 \times 10^{-5}$	2.38	1.00	$0.607 \times 10^{-3}$		
80	47.39	971.8	0.2935	2309	4197	1962	0.670	0.0230	$0.355 \times 10^{-3}$	$1.159 \times 10^{-5}$	2.22	1.00	$0.653 \times 10^{-3}$		
85	57.83	968.1	0.3536	2296	4201	1977	0.673	0.0235	$0.333 \times 10^{-3}$	$1.176 \times 10^{-5}$	2.08	1.00	$0.670 \times 10^{-3}$		
90	70.14	965.3	0.4235	2283	4206	1993	0.675	0.0240	$0.315 \times 10^{-3}$	$1.193 \times 10^{-5}$	1.96	1.00	$0.702 \times 10^{-3}$		
95	84.55	961.5	0.5045	2270	4212	2010	0.677	0.0246	$0.297 \times 10^{-3}$	$1.210 \times 10^{-5}$	1.85	1.00	$0.716 \times 10^{-3}$		
100	101.33	957.9	0.5978	2257	4217	2029	0.679	0.0251	$0.282 \times 10^{-3}$	$1.227 \times 10^{-5}$	1.75	1.00	$0.750 \times 10^{-3}$		
110	143.27	950.6	0.8263	2230	4229	2071	0.682	0.0262	$0.255 \times 10^{-3}$	$1.261 \times 10^{-5}$	1.58	1.00	$0.798 \times 10^{-3}$		
120	198.53	943.4	1.121	2203	4244	2120	0.683	0.0275	$0.232 \times 10^{-3}$	$1.296 \times 10^{-5}$	1.44	1.00	$0.858 \times 10^{-3}$		
130	270.1	934.6	1.496	2174	4263	2177	0.684	0.0288	$0.213 \times 10^{-3}$	$1.330 \times 10^{-5}$	1.33	1.01	$0.913 \times 10^{-3}$		
140	361.3	921.7	1.965	2145	4286	2244	0.683	0.0301	$0.197 \times 10^{-3}$	$1.365 \times 10^{-5}$	1.24	1.02	$0.970 \times 10^{-3}$		
150	475.8	916.6	2.546	2114	4311	2314	0.682	0.0316	$0.183 \times 10^{-3}$	$1.399 \times 10^{-5}$	1.16	1.02	$1.025 \times 10^{-3}$		
160	617.8	907.4	3.256	2083	4340	2420	0.680	0.0331	$0.170 \times 10^{-3}$	$1.434 \times 10^{-5}$	1.09	1.05	$1.145 \times 10^{-3}$		
170	791.7	897.7	4.119	2050	4370	2490	0.677	0.0347	$0.160 \times 10^{-3}$	$1.468 \times 10^{-5}$	1.03	1.05	$1.178 \times 10^{-3}$		
180	1,002.1	887.3	5.153	2015	4410	2590	0.673	0.0364	$0.150 \times 10^{-3}$	$1.502 \times 10^{-5}$	0.983	1.07	$1.210 \times 10^{-3}$		
190	1,254.4	876.4	6.388	1979	4460	2710	0.669	0.0382	$0.142 \times 10^{-3}$	$1.537 \times 10^{-5}$	0.947	1.09	$1.280 \times 10^{-3}$		
200	1,553.8	864.3	7.852	1941	4500	2840	0.663	0.0401	$0.134 \times 10^{-3}$	$1.571 \times 10^{-5}$	0.910	1.11	$1.350 \times 10^{-3}$		
220	2,318	840.3	11.60	1859	4610	3110	0.650	0.0442	$0.122 \times 10^{-3}$	$1.641 \times 10^{-5}$	0.865	1.15	$1.520 \times 10^{-3}$		
240	3,344	813.7	16.73	1767	4760	3520	0.632	0.0487	$0.111 \times 10^{-3}$	$1.712 \times 10^{-5}$	0.836	1.24	$1.720 \times 10^{-3}$		
260	4,688	783.7	23.69	1663	4970	4070	0.609	0.0540	$0.102 \times 10^{-3}$	$1.788 \times 10^{-5}$	0.832	1.35	$2.000 \times 10^{-3}$		
280	6,412	750.8	33.15	1544	5280	4835	0.581	0.0605	$0.094 \times 10^{-3}$	$1.870 \times 10^{-5}$	0.854	1.49	$2.380 \times 10^{-3}$		
300	8,581	713.8	46.15	1405	5750	5980	0.548	0.0695	$0.086 \times 10^{-3}$	$1.965 \times 10^{-5}$	0.902	1.69	$2.950 \times 10^{-3}$		
30	11,274	667.1	64.57	1239	6540	7900	0.509	0.0836	$0.078 \times 10^{-3}$	$2.084 \times 10^{-5}$	1.00	1.97			
+40	14,586	610.5	92.62	1028	8240	11,870	0.469	0.110	$0.070 \times 10^{-3}$	$2.255 \times 10^{-5}$	1.23	2.43			
360	18,651	528.3	144.0	720	14,690	25,800	0.427	0.178	$0.060 \times 10^{-3}$	$2.571 \times 10^{-5}$	2.06	3.73			
374.14	22,090	317.0	317.0	0	—	—	—	—	$0.043 \times 10^{-3}$	$4.313 \times 10^{-5}$					

Note 1: Kinematic viscosity  $\nu$  and thermal diffusivity  $\alpha$  can be calculated from their definitions,  $\nu = \mu/\rho$  and  $\alpha = k/\rho c_p = \nu/\text{Pr}$ . The temperatures 0.01°C, 100°C, and 374.14°C are the triple-, boiling-, and critical-point temperatures of water, respectively. The properties listed above (except the vapor density) can be used at any pressure with negligible error except at temperatures near the critical-point value.

Note 2: The unit kJ/kg·°C for specific heat is equivalent to J/kg·K, and the unit W/m·°C for thermal conductivity is equivalent to W/m·K.

Source: Viscosity and thermal conductivity data are from J. V. Sengers and J. T. R. Watson, *Journal of Physical and Chemical Reference Data* 15 (1986), pp. 1291–1322. Other data are obtained from various sources or calculated.

TABLE A-13

Properties of liquids

Temp. <i>T</i> , °C	Density <i>ρ</i> , kg/m <sup>3</sup>	Specific Heat <i>c<sub>p</sub></i> , J/kg·K	Thermal Conductivity <i>k</i> , W/m·K	Thermal Diffusivity <i>α</i> , m <sup>2</sup> /s	Dynamic Viscosity <i>μ</i> , kg/m·s	Kinematic Viscosity <i>ν</i> , m <sup>2</sup> /s	Prandtl Number Pr	Volume Expansion Coeff. <i>β</i> , 1/K
<i>Methane [CH<sub>4</sub>]</i>								
-160	420.2	3492	0.1863	$1.270 \times 10^{-7}$	$1.133 \times 10^{-4}$	$2.699 \times 10^{-7}$	2.126	0.00352
-150	405.0	3580	0.1703	$1.174 \times 10^{-7}$	$9.169 \times 10^{-5}$	$2.264 \times 10^{-7}$	1.927	0.00391
-140	388.8	3700	0.1550	$1.077 \times 10^{-7}$	$7.551 \times 10^{-5}$	$1.942 \times 10^{-7}$	1.803	0.00444
-130	371.1	3875	0.1402	$9.749 \times 10^{-8}$	$6.288 \times 10^{-5}$	$1.694 \times 10^{-7}$	1.738	0.00520
-120	351.4	4146	0.1258	$8.634 \times 10^{-8}$	$5.257 \times 10^{-5}$	$1.496 \times 10^{-7}$	1.732	0.00637
-110	328.8	4611	0.1115	$7.356 \times 10^{-8}$	$4.377 \times 10^{-5}$	$1.331 \times 10^{-7}$	1.810	0.00841
-100	301.0	5578	0.0967	$5.761 \times 10^{-8}$	$3.577 \times 10^{-5}$	$1.188 \times 10^{-7}$	2.063	0.01282
-90	261.7	8902	0.0797	$3.423 \times 10^{-8}$	$2.761 \times 10^{-5}$	$1.055 \times 10^{-7}$	3.082	0.02922
<i>Methanol [CH<sub>3</sub>(OH)]</i>								
20	788.4	2515	0.1987	$1.002 \times 10^{-7}$	$5.857 \times 10^{-4}$	$7.429 \times 10^{-7}$	7.414	0.00118
30	779.1	2577	0.1980	$9.862 \times 10^{-8}$	$5.088 \times 10^{-4}$	$6.531 \times 10^{-7}$	6.622	0.00120
40	769.6	2644	0.1972	$9.690 \times 10^{-8}$	$4.460 \times 10^{-4}$	$5.795 \times 10^{-7}$	5.980	0.00123
50	760.1	2718	0.1965	$9.509 \times 10^{-8}$	$3.942 \times 10^{-4}$	$5.185 \times 10^{-7}$	5.453	0.00127
60	750.4	2798	0.1957	$9.320 \times 10^{-8}$	$3.510 \times 10^{-4}$	$4.677 \times 10^{-7}$	5.018	0.00132
70	740.4	2885	0.1950	$9.128 \times 10^{-8}$	$3.146 \times 10^{-4}$	$4.250 \times 10^{-7}$	4.655	0.00137
<i>Isobutane (R600a)</i>								
-100	683.8	1881	0.1383	$1.075 \times 10^{-7}$	$9.305 \times 10^{-4}$	$1.360 \times 10^{-6}$	12.65	0.00142
-75	659.3	1970	0.1357	$1.044 \times 10^{-7}$	$5.624 \times 10^{-4}$	$8.531 \times 10^{-7}$	8.167	0.00150
-50	634.3	2069	0.1283	$9.773 \times 10^{-8}$	$3.769 \times 10^{-4}$	$5.942 \times 10^{-7}$	6.079	0.00161
-25	608.2	2180	0.1181	$8.906 \times 10^{-8}$	$2.688 \times 10^{-4}$	$4.420 \times 10^{-7}$	4.963	0.00177
0	580.6	2306	0.1068	$7.974 \times 10^{-8}$	$1.993 \times 10^{-4}$	$3.432 \times 10^{-7}$	4.304	0.00199
25	550.7	2455	0.0956	$7.069 \times 10^{-8}$	$1.510 \times 10^{-4}$	$2.743 \times 10^{-7}$	3.880	0.00232
50	517.3	2640	0.0851	$6.233 \times 10^{-8}$	$1.155 \times 10^{-4}$	$2.233 \times 10^{-7}$	3.582	0.00286
75	478.5	2896	0.0757	$5.460 \times 10^{-8}$	$8.785 \times 10^{-5}$	$1.836 \times 10^{-7}$	3.363	0.00385
100	429.6	3361	0.0669	$4.634 \times 10^{-8}$	$6.483 \times 10^{-5}$	$1.509 \times 10^{-7}$	3.256	0.00628
<i>Glycerin</i>								
0	1276	2262	0.2820	$9.773 \times 10^{-8}$	10.49	$8.219 \times 10^{-3}$	84,101	
5	1273	2288	0.2835	$9.732 \times 10^{-8}$	6.730	$5.287 \times 10^{-3}$	54,327	
10	1270	2320	0.2846	$9.662 \times 10^{-8}$	4.241	$3.339 \times 10^{-3}$	34,561	
15	1267	2354	0.2856	$9.576 \times 10^{-8}$	2.496	$1.970 \times 10^{-3}$	20,570	
20	1264	2386	0.2860	$9.484 \times 10^{-8}$	1.519	$1.201 \times 10^{-3}$	12,671	
25	1261	2416	0.2860	$9.388 \times 10^{-8}$	0.9934	$7.878 \times 10^{-4}$	8,392	
30	1258	2447	0.2860	$9.291 \times 10^{-8}$	0.6582	$5.232 \times 10^{-4}$	5,631	
35	1255	2478	0.2860	$9.195 \times 10^{-8}$	0.4347	$3.464 \times 10^{-4}$	3,767	
40	1252	2513	0.2863	$9.101 \times 10^{-8}$	0.3073	$2.455 \times 10^{-4}$	2,697	
<i>Engine Oil (unused)</i>								
0	899.0	1797	0.1469	$9.097 \times 10^{-8}$	3.814	$4.242 \times 10^{-3}$	46,636	0.00070
20	888.1	1881	0.1450	$8.680 \times 10^{-8}$	0.8374	$9.429 \times 10^{-4}$	10,863	0.00070
40	876.0	1964	0.1444	$8.391 \times 10^{-8}$	0.2177	$2.485 \times 10^{-4}$	2,962	0.00070
60	863.9	2048	0.1404	$7.934 \times 10^{-8}$	0.07399	$8.565 \times 10^{-5}$	1,080	0.00070
80	852.0	2132	0.1380	$7.599 \times 10^{-8}$	0.03232	$3.794 \times 10^{-5}$	499.3	0.00070
100	840.0	2220	0.1367	$7.330 \times 10^{-8}$	0.01718	$2.046 \times 10^{-5}$	279.1	0.00070
120	828.9	2308	0.1347	$7.042 \times 10^{-8}$	0.01029	$1.241 \times 10^{-5}$	176.3	0.00070
140	816.8	2395	0.1330	$6.798 \times 10^{-8}$	0.006558	$8.029 \times 10^{-6}$	118.1	0.00070
150	810.3	2441	0.1327	$6.708 \times 10^{-8}$	0.005344	$6.595 \times 10^{-6}$	98.31	0.00070

Source: Data generated from the EES software developed by S. A. Klein and F. L. Alvarado. Originally based on various sources.

TABLE A-15

Properties of air at 1 atm pressure

Temp. <i>T</i> , °C	Density <i>ρ</i> , kg/m <sup>3</sup>	Specific Heat <i>c<sub>p</sub></i> , J/kg·K	Thermal Conductivity <i>k</i> , W/m·K	Thermal Diffusivity <i>α</i> , m <sup>2</sup> /s	Dynamic Viscosity <i>μ</i> , kg/m·s	Kinematic Viscosity <i>ν</i> , m <sup>2</sup> /s	Prandtl Number Pr
-150	2.866	983	0.01171	$4.158 \times 10^{-6}$	$8.636 \times 10^{-6}$	$3.013 \times 10^{-6}$	0.7246
-100	2.038	966	0.01582	$8.036 \times 10^{-6}$	$1.189 \times 10^{-5}$	$5.837 \times 10^{-6}$	0.7263
-50	1.582	999	0.01979	$1.252 \times 10^{-5}$	$1.474 \times 10^{-5}$	$9.319 \times 10^{-6}$	0.7440
-40	1.514	1002	0.02057	$1.356 \times 10^{-5}$	$1.527 \times 10^{-5}$	$1.008 \times 10^{-5}$	0.7436
-30	1.451	1004	0.02134	$1.465 \times 10^{-5}$	$1.579 \times 10^{-5}$	$1.087 \times 10^{-5}$	0.7425
-20	1.394	1005	0.02211	$1.578 \times 10^{-5}$	$1.630 \times 10^{-5}$	$1.169 \times 10^{-5}$	0.7408
-10	1.341	1006	0.02288	$1.696 \times 10^{-5}$	$1.680 \times 10^{-5}$	$1.252 \times 10^{-5}$	0.7387
0	1.292	1006	0.02364	$1.818 \times 10^{-5}$	$1.729 \times 10^{-5}$	$1.338 \times 10^{-5}$	0.7362
5	1.269	1006	0.02401	$1.880 \times 10^{-5}$	$1.754 \times 10^{-5}$	$1.382 \times 10^{-5}$	0.7350
10	1.246	1006	0.02439	$1.944 \times 10^{-5}$	$1.778 \times 10^{-5}$	$1.426 \times 10^{-5}$	0.7336
15	1.225	1007	0.02476	$2.009 \times 10^{-5}$	$1.802 \times 10^{-5}$	$1.470 \times 10^{-5}$	0.7323
20	1.204	1007	0.02514	$2.074 \times 10^{-5}$	$1.825 \times 10^{-5}$	$1.516 \times 10^{-5}$	0.7309
25	1.184	1007	0.02551	$2.141 \times 10^{-5}$	$1.849 \times 10^{-5}$	$1.562 \times 10^{-5}$	0.7296
30	1.164	1007	0.02588	$2.208 \times 10^{-5}$	$1.872 \times 10^{-5}$	$1.608 \times 10^{-5}$	0.7282
35	1.145	1007	0.02625	$2.277 \times 10^{-5}$	$1.895 \times 10^{-5}$	$1.655 \times 10^{-5}$	0.7268
40	1.127	1007	0.02662	$2.346 \times 10^{-5}$	$1.918 \times 10^{-5}$	$1.702 \times 10^{-5}$	0.7255
45	1.109	1007	0.02699	$2.416 \times 10^{-5}$	$1.941 \times 10^{-5}$	$1.750 \times 10^{-5}$	0.7241
50	1.092	1007	0.02735	$2.487 \times 10^{-5}$	$1.963 \times 10^{-5}$	$1.798 \times 10^{-5}$	0.7228
60	1.059	1007	0.02808	$2.632 \times 10^{-5}$	$2.008 \times 10^{-5}$	$1.896 \times 10^{-5}$	0.7202
70	1.028	1007	0.02881	$2.780 \times 10^{-5}$	$2.052 \times 10^{-5}$	$1.995 \times 10^{-5}$	0.7177
80	0.9994	1008	0.02953	$2.931 \times 10^{-5}$	$2.096 \times 10^{-5}$	$2.097 \times 10^{-5}$	0.7154
90	0.9718	1008	0.03024	$3.086 \times 10^{-5}$	$2.139 \times 10^{-5}$	$2.201 \times 10^{-5}$	0.7132
100	0.9458	1009	0.03095	$3.243 \times 10^{-5}$	$2.181 \times 10^{-5}$	$2.306 \times 10^{-5}$	0.7111
120	0.8977	1011	0.03235	$3.565 \times 10^{-5}$	$2.264 \times 10^{-5}$	$2.522 \times 10^{-5}$	0.7073
140	0.8542	1013	0.03374	$3.898 \times 10^{-5}$	$2.345 \times 10^{-5}$	$2.745 \times 10^{-5}$	0.7041
160	0.8148	1016	0.03511	$4.241 \times 10^{-5}$	$2.420 \times 10^{-5}$	$2.975 \times 10^{-5}$	0.7014
180	0.7788	1019	0.03646	$4.593 \times 10^{-5}$	$2.504 \times 10^{-5}$	$3.212 \times 10^{-5}$	0.6992
200	0.7459	1023	0.03779	$4.954 \times 10^{-5}$	$2.577 \times 10^{-5}$	$3.455 \times 10^{-5}$	0.6974
250	0.6746	1033	0.04104	$5.890 \times 10^{-5}$	$2.760 \times 10^{-5}$	$4.091 \times 10^{-5}$	0.6946
300	0.6158	1044	0.04418	$6.871 \times 10^{-5}$	$2.934 \times 10^{-5}$	$4.765 \times 10^{-5}$	0.6935
350	0.5664	1056	0.04721	$7.892 \times 10^{-5}$	$3.101 \times 10^{-5}$	$5.475 \times 10^{-5}$	0.6937
400	0.5243	1069	0.05015	$8.951 \times 10^{-5}$	$3.261 \times 10^{-5}$	$6.219 \times 10^{-5}$	0.6948
450	0.4880	1081	0.05298	$1.004 \times 10^{-4}$	$3.415 \times 10^{-5}$	$6.997 \times 10^{-5}$	0.6965
500	0.4565	1093	0.05572	$1.117 \times 10^{-4}$	$3.563 \times 10^{-5}$	$7.806 \times 10^{-5}$	0.6986
600	0.4042	1115	0.06093	$1.352 \times 10^{-4}$	$3.846 \times 10^{-5}$	$9.515 \times 10^{-5}$	0.7037
700	0.3627	1135	0.06581	$1.598 \times 10^{-4}$	$4.111 \times 10^{-5}$	$1.133 \times 10^{-4}$	0.7092
800	0.3289	1153	0.07037	$1.855 \times 10^{-4}$	$4.362 \times 10^{-5}$	$1.326 \times 10^{-4}$	0.7149
900	0.3008	1169	0.07465	$2.122 \times 10^{-4}$	$4.600 \times 10^{-5}$	$1.529 \times 10^{-4}$	0.7206
1000	0.2772	1184	0.07868	$2.398 \times 10^{-4}$	$4.826 \times 10^{-5}$	$1.741 \times 10^{-4}$	0.7260
1500	0.1990	1234	0.09599	$3.908 \times 10^{-4}$	$5.817 \times 10^{-5}$	$2.922 \times 10^{-4}$	0.7478
2000	0.1553	1264	0.11113	$5.664 \times 10^{-4}$	$6.630 \times 10^{-5}$	$4.270 \times 10^{-4}$	0.7539

Note: For ideal gases, the properties  $c_p$ ,  $k$ ,  $\mu$ , and Pr are independent of pressure. The properties  $\rho$ ,  $\nu$ , and  $\alpha$  at a pressure  $P$  (in atm) other than 1 atm are determined by multiplying the values of  $\rho$  at the given temperature by  $P$  and by dividing  $\nu$  and  $\alpha$  by  $P$ .

Source: Data generated from the EES software developed by S. A. Klein and F. L. Alvarado. Original sources: Keenan, Chao, Keyes, Gas Tables, Wiley, 1984; and Thermophysical Properties of Matter, Vol. 3: Thermal Conductivity, Y. S. Touloukian, P. E. Liley, S. C. Saxena, Vol. 11: Viscosity, Y. S. Touloukian, S. C. Saxena, and P. Hestermanns, IFI/Plenum, NY, 1970, ISBN 0-306067020-8.

